

REACTION OF 2,3-DICHLOROBENZO[b]THIOPHENE  
1,1-DIOXIDE WITH 3-AMINOPROPYLSILANES, 3-  
AMINOPROPYLSILOXANES, AND 3-AMINOPROPYLSILATRANE

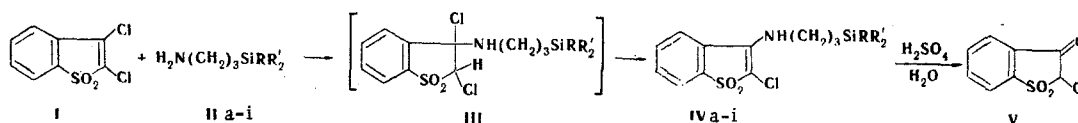
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2-Chloro-3-[ $\gamma$ -(organosilyl)propylamine]benzo[b]thiophene 1,1-dioxides are formed in the reaction of 2,3-dichlorobenzo[b]thiophene 1,1-dioxide with 3-aminopropylsilanes, 3-aminopropylsiloxanes, and 3-aminopropylsilatrane.

Continuing our investigation of nucleophilic addition to 2,3-dichlorobenzo[b]thiophene 1,1-dioxide (I) [1], which is an effective fungicide [2, 3], we decided to study the reaction of I with 3-aminopropylsilanes, which display antimicrobial activity [4, 5], in order to obtain organosilicon derivatives of benzo[b]thiophene and to determine the fungistatic activity of the compounds obtained.

When I is refluxed with trialkyl(3-aminopropyl)silanes, heptamethyl-3-( $\gamma$ -aminopropyl)trisiloxane, 3-aminopropyltriethoxysilane, and 1-( $\gamma$ -aminopropyl)silatrane (in a molar ratio of 1:3) in benzene for 3 h, 2-chloro-3-[ $\gamma$ -(organosilyl)propylamino]benzo[b]thiophene 1,1-dioxides (IV) and the hydrochlorides of the corresponding 3-aminopropylsilanes or 3-aminopropylsilatrane (Table 1) are formed in yields of 60-70%:



We were unable to obtain the assumed intermediate (III). Lowering of the temperature from 80 to 20°C and changing the molar ratio of the reagents to 1:1 led only to a considerable decrease in the yield of IV. Addition of the aminoalkylsilanes and dehydrochlorination apparently proceed simultaneously.

The structure of IV was established by their acid hydrolysis to 3-oxo-2-chloro-2,3-dihydrobenzo[b]thiophene 1,1-dioxide (V) and was confirmed by the IR spectra, which contain the following absorption bands:  $\nu_{\text{CN}}$  at 1275-1285  $\text{cm}^{-1}$ ,  $\nu_{\text{NH}}$  at 3360-3390  $\text{cm}^{-1}$ ,  $\nu_{\text{SO}_2}$  at 1335-1340 and 1160  $\text{cm}^{-1}$ , and  $\delta_{\text{CHSi}}$  at 1260-1270  $\text{cm}^{-1}$ . The absorption band at 578  $\text{cm}^{-1}$  for IVi can be assigned to the stretching vibrations of the Si ← N bond [6].

Compound V was also formed in an attempt to obtain the hydrochlorides of IV by the action of alcoholic hydrogen chloride.

Compound I and all of the aminoalkylsilanes (IIa-h) used suppress the growth of the pathogenic fungi *Candida albicans* 17/846, *Epidermophyton Kaufmann-Wolf* 41, and *Trichophyton gypseum* 4/3. The minimum concentration that suppresses the growth of the fungi is 55.6, 3.5, and 6.9  $\mu\text{g}/\text{ml}$  for I, respectively, and 7.8, 15.6, and 7.8  $\mu\text{g}/\text{ml}$  for diethylamyl derivative IIg. However, the introduction of an aminopropylsilyl grouping (in place of the 3-aminopropyltriethoxysilyl grouping) into the I molecule deprives it of its fungistatic activity.

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TABLE 1. 2-Chloro-3-[ $\gamma$ -(organosilyl)propylamino]benzo[b]-thiophene 1,1-Dioxides (IVa-i)

Com- pound	R	R'	mp, °C	Empirical formula	Found, %			Calc., %			
					C	H	N	C	H	N	
IVa	CH <sub>3</sub>	C <sub>6</sub> H <sub>9</sub>	149	C <sub>20</sub> H <sub>32</sub> ClNO <sub>2</sub> SSi	57.8	8.0	3.4	58.0	7.8	3.4	57
IVb	CH <sub>3</sub>	(CH <sub>3</sub> ) <sub>3</sub> SiO	152	C <sub>18</sub> H <sub>32</sub> ClNO <sub>2</sub> SSi <sub>3</sub>	46.5	6.9	3.0	46.7	6.7	2.9	58
IVc	C <sub>2</sub> H <sub>5</sub>	C <sub>5</sub> H <sub>11</sub>	129	C <sub>23</sub> H <sub>38</sub> ClNO <sub>2</sub> SSi	60.8	8.6	3.1	60.6	8.4	3.1	68
IVd	C <sub>2</sub> H <sub>5</sub>	C <sub>2</sub> H <sub>5</sub>	178	C <sub>17</sub> H <sub>26</sub> ClNO <sub>2</sub> SSi	54.9	7.3	3.8	54.9	7.1	3.8	73
IVf	C <sub>3</sub> H <sub>7</sub>	C <sub>2</sub> H <sub>5</sub>	172	C <sub>18</sub> H <sub>28</sub> ClNO <sub>2</sub> SSi	56.3	7.4	3.5	56.0	7.3	3.6	70
IVg	C <sub>4</sub> H <sub>9</sub>	C <sub>2</sub> H <sub>5</sub>	163	C <sub>19</sub> H <sub>30</sub> ClNO <sub>2</sub> SSi	56.5	7.5	3.5	57.1	7.6	3.5	70
IVh	C <sub>5</sub> H <sub>11</sub>	C <sub>2</sub> H <sub>5</sub>	145	C <sub>20</sub> H <sub>32</sub> ClNO <sub>2</sub> SSi	58.0	7.9	3.0	58.0	7.8	3.3	68
IVi	C <sub>2</sub> H <sub>5</sub> O	C <sub>5</sub> H <sub>5</sub> O	137	C <sub>17</sub> H <sub>26</sub> ClNO <sub>2</sub> SSi	48.8	6.3	3.2	48.6	6.2	3.3	56
IVj	(OCH <sub>2</sub> CH <sub>2</sub> ) <sub>3</sub> N		226-228	C <sub>17</sub> H <sub>23</sub> ClN <sub>2</sub> O <sub>5</sub> SSi	47.6	5.3	6.6	47.4	5.4	6.5	79

## EXPERIMENTAL

2-Chloro-3-[ $\gamma$ -(organosilyl)propylamino]benzo[b]thiophene 1,1-Dioxides (IVa-h) (see Table 1). A solution of 4 mmole of I [7] and 12 mmole of II in 15 ml of dry benzene was refluxed for 3 h, after which it was filtered and the solvent was partially removed by distillation. The precipitated IV was washed with hexane and recrystallized from benzene-hexane.

2-Chloro-3-[ $\gamma$ -(1-silatranlyl)propylamino]benzo[b]thiophene 1,1-Dioxide (IVi). A solution of 4 mmole of I and 12 mmole of III [8] in 20 ml of dry benzene was refluxed for 3 h, after which the precipitate was removed by filtration, washed with water, and recrystallized from dimethylformamide.

3-Oxo-2-chloro-2,3-dihydrobenzo[b]thiophene 1,1-Dioxide. This compound was obtained by hydrolysis of IV with 6 N H<sub>2</sub>SO<sub>4</sub> [1] and had mp 154-155° and  $\nu$ CO 1740 cm<sup>-1</sup>.

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